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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/720,617	11/24/2003	David M. Lowe	2003B125	4251
23455	7590	04/12/2006	EXAMINER	
EXXONMOBIL CHEMICAL COMPANY 5200 BAYWAY DRIVE P.O. BOX 2149 BAYTOWN, TX 77522-2149			HAILEY, PATRICIA L	
			ART UNIT	PAPER NUMBER
			1755	

DATE MAILED: 04/12/2006

Please find below and/or attached an Office communication concerning this application or proceeding.

<b>Office Action Summary</b>	<b>Application No.</b>	<b>Applicant(s)</b>	
	10/720,617	LOWE ET AL.	
	Examiner Patricia L. Hailey	Art Unit 1755	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --  
**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

#### Status

- 1) Responsive to communication(s) filed on 30 January 2006.
- 2a) This action is FINAL.                    2b) This action is non-final.
- 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

#### Disposition of Claims

- 4) Claim(s) 1-15 and 19-31 is/are pending in the application.
- 4a) Of the above claim(s) 22-30 is/are withdrawn from consideration.
- 5) Claim(s) \_\_\_\_\_ is/are allowed.
- 6) Claim(s) 1-15, 19-21, and 31 is/are rejected.
- 7) Claim(s) \_\_\_\_\_ is/are objected to.
- 8) Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

#### Application Papers

- 9) The specification is objected to by the Examiner.
- 10) The drawing(s) filed on \_\_\_\_\_ is/are: a) accepted or b) objected to by the Examiner.  
 Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
 Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

#### Priority under 35 U.S.C. § 119

- 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) All    b) Some \* c) None of:
1. Certified copies of the priority documents have been received.
  2. Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
  3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

#### Attachment(s)

- |  |   |
|--|---|
| 1) <input type="checkbox"/> Notice of References Cited (PTO-892)   | 4) <input type="checkbox"/> Interview Summary (PTO-413)                     |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948)                                     | Paper No(s)/Mail Date. _____ .  |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)<br>Paper No(s)/Mail Date _____ . | 5) <input type="checkbox"/> Notice of Informal Patent Application (PTO-152) |
|  | 6) <input type="checkbox"/> Other: _____ .                                  |

Applicants' remarks and amendments, filed on January 30, 2006, have been carefully considered. Claims 16-18 have been canceled; no new claims have been added.

Claims 1-15 and 19-31 remain pending in this application.

*Election/Restrictions*

1. Claims 22-30 remain withdrawn from further consideration pursuant to 37 CFR 1.142(b), as being drawn to a nonelected process for selectively removing C<sub>2</sub> to C<sub>4</sub> alkynes or diolefins from a feedstock also containing olefins, there being no allowable generic or linking claim. Applicant timely traversed the restriction (election) requirement in the reply filed on October 14, 2004.

Claims 1-21 and 31 remain under consideration by the Examiner.

*Withdrawn Rejections*

The 112(2) rejection of claims 16-18 stated in the previous Office Action has been withdrawn in view of Applicants' cancellation of these claims.

*Maintained Rejections*

2. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

The following rejections of record are maintained:

***Claim Rejections - 35 USC § 103***

3. ***Claims 1-10 stand rejected under 35 U.S.C. 103(a) as being unpatentable over Moser et al. (U. S. Patent No. 6,514,904).***

Moser et al. teach an alumina (col. 1, line 3 to col. 2, line 48) catalyst with a platinum group metal component (e.g., rhodium), incorporated therein via coprecipitation, ion exchange, or impregnation, using salts such as rhodium nitrate (col. 5, lines 39-40). See col. 5, lines 3-57 of Moser et al., especially lines 14-20, which discloses a platinum group component mass percentage range of about 0.01 to about 2%.

Moser et al. also disclose that the catalyst may contain optional modifiers such as indium, in amounts ranging from about 0.01 to about 5 mass %. See col. 6, lines 55-67 of Moser et al., which also discloses that the optional modifiers may be incorporated either during or after the preparation of the carrier material (i.e., alumina), or before, during or after the incorporation of the other catalyst components.

Because Moser et al. teach mass percentage ranges that read upon those respectively claimed, one of ordinary skill in the art would easily determine via known mathematic techniques that the molar ratios of platinum group component (e.g., rhodium), to modifier (e.g., indium) would read upon that respectively claimed.

In addition to the aforementioned preparation techniques, calcination and reduction steps are also employed. Calcination can be performed preferably before

incorporation of any metals into the support, but also can be performed at temperatures ranging from about 370°C to about 600°C. See col. 7, lines 16-39 of Moser et al.

The reduction step can be performed in a substantially water-free environment, and in the presence of gases such as hydrogen, nitrogen, or carbon monoxide, at temperatures ranging from about 315°C to about 650°C. See col. 8, lines 6-30 of Moser et al.

Moser et al. do not teach the exact percentage ranges for rhodium and indium, as recited in the instant claims. However, the percentage ranges taught by Moser et al. either are within or overlap the percentage ranges respectively recited in the instant claims.

The subject matter as a whole would have been obvious to one having ordinary skill in the art at the time the invention was made to have selected the overlapping portion of the range disclosed by the reference because overlapping ranges have been held to be a *prima facie* case of obviousness. In re Malagari, 182 U.S.P.Q. 549.

Further, where claimed ranges "overlap or lie inside ranges disclosed by the prior art a *prima facie* case of obviousness exists." In re Wertheim, 541 F.2d 257, 191 U.S.P.Q. 90 (CCPA 1976).

**4. *Claims 1-15, 19-21, and 31 stand rejected under 35 U.S.C. 103(a) as being unpatentable over Robinson et al. (U. S. Patent No. 4,522,935).***

Robinson et al. teach a catalyst comprising a platinum group component, an indium component, and a porous support material, wherein the atomic ratio of indium

to platinum group component is more than about 1.14. See col. 2, lines 26-35 of Robinson et al. Although this ratio is the reverse of that recited in Applicants' claims, one skilled in the art would easily deduce that inverting the atomic ratio of Robinson et al. would result in a ratio comparable to that claimed by Applicants.

The platinum group component is present in an amount ranging from about 0.01 to about 5 wt. %, and the indium component is present in an amount ranging from about 0.01 to about 15 wt. %. See col. 3, lines 5-14 of Robinson et al. Examples of the platinum group component include rhodium. See col. 4, lines 20-34 of Robinson et al.

Both the platinum group component and the indium component may be incorporated into the support material via cogelation or coprecipitation with the support material, or by ion exchange or impregnation of the support material. With the indium component, compounds such as indium chloride or indium nitrate may be employed as impregnating solutions. See col. 4, lines 35-52 and col. 5, lines 10-46 of Robinson et al.

Examples of the support material include refractory oxides such as alumina and zirconium dioxide (zirconia). See col. 6, line 4 to col. 7, line 2 of Robinson et al.

In addition to the aforementioned methods of combining the catalyst components with the support material, techniques such as calcination and reduction are employed. Calcination temperatures range from about 700°F to about 1100°F (371.1°C to 593.3°C); reduction is performed under dry hydrogen at conditions including a

temperature of about 400°F to about 1200°F (204.4°C to 648.8°C). See col. 7, line 50 to col. 8, line 19 of Robinson et al.

Robinson et al. do not teach the exact percentage ranges for rhodium and indium, as recited in the instant claims. However, the percentage ranges taught by Robinson et al. either are within or overlap the percentage ranges respectively recited in the instant claims.

The subject matter as a whole would have been obvious to one having ordinary skill in the art at the time the invention was made to have selected the overlapping portion of the range disclosed by the reference because overlapping ranges have been held to be a *prima facie* case of obviousness. In re Malagari, 182 U.S.P.Q. 549.

Further, where claimed ranges “overlap or lie inside ranges disclosed by the prior art a *prima facie* case of obviousness exists.” In re Wertheim, 541 F.2d 257, 191 U.S.P.Q. 90 (CCPA 1976).

5. *Claims 11-15 and 19-21 stand rejected under 35 U.S.C. 103(a) as being unpatentable over Moser et al. (U. S. Patent No. 6,514,904) in view of Robinson et al. (U. S. Patent No. 4,522,935).*

Both Moser et al. and Robinson et al. are relied upon for their teachings with respect to claims 1-10 and 31. However, while Moser et al. disclose the employment of rhodium nitrate (col. 5, lines 39-40) in preparing Patentees’ catalyst, as well as the feasibility in employing optional modifiers such as indium (col. 6, lines 55-67), Moser et

al. do not disclose the employment of indium formate or indium nitrate as sources for the indium compound.

Robinson et al. is relied upon to disclose the employment of indium nitrate. See col. 4, lines 35-52 and col. 5, lines 10-46 of Robinson et al.

Because the catalysts of both Moser et al. and Robinson et al. are suitable for catalytic reforming of gasoline-range hydrocarbons, as well as other processes such as alkylation, dealkylation, transalkylation, etc., motivation to combine the references is deemed proper. See col. 8, lines 42-47 of Moser et al., and col. 8, lines 33-35 and col. 9, lines 27-32 of Robinson et al.

It would have been obvious to one skilled in the art at the time the invention was made to modify the teachings of Moser et al. by incorporating therein indium nitrate as a source for indium as a modifier, because combining two or more materials disclosed by the prior art for the same purpose to form a third material that is to be used for the same purpose has been held to be a *prima facie* case of obviousness, see In re Kerkhoven, 205 U.S.P.Q. 1069.

#### *Response to Arguments*

Although the claims in their present form are directed to a catalyst "consisting essentially of" a rhodium component, and an indium component, it is agreed that such language limits the scope of a claim to the ingredients specified therein, as well as those not materially affecting the **basic and novel characteristics** of said composition.

However, the claims in their present form do not recite any of said basic and novel characteristics; Applicants' claims are merely directed to a "catalyst composition". Although the cited references "are directed to a catalyst for reforming hydrocarbons", the references continue to read upon Applicants' claims regarding the claimed components and their respective amounts.

Further, Moser et al. at col. 8, lines 34-47 disclose additional utilities for Patentees' catalyst including catalytic reforming (i.e., "other hydrocarbon conversions", line 47), as does Robinson et al. at col. 9, lines 11-32, which discloses that the "catalyst of this invention may be utilized for catalyzing many reactions besides the reforming reactions discussed above, including...., hydrogenation,..." (lines 27-31). From these teachings, one of ordinary skill in the art would find reasonable expectation that these prior art catalysts could and would function in the selective hydrogenation of alkynes and diolefins to olefins, absent the showing of convincing evidence to the contrary.

Applicants' addition to claim 31 of the phrase "effective for the selective hydrogenation of alkynes and diolefins to olefins" is noted, but is not considered to further distinguish the claim from the prior art teachings. Such a limitation is not given patentable weight, because, in terms of the claimed catalyst and its components, the prior art references continue to read upon Applicants' claims.

It is well settled that when a claimed composition appears to be substantially the same as a composition disclosed in the prior art, the burden is properly upon the applicant to prove by way of tangible evidence that the prior art composition does not

necessarily possess characteristics attributed to the CLAIMED composition. In re Spada, 911 F.2d 705, 15 USPQ2d 1655 (Fed. Circ. 1990); In re Fitzgerald, 619 F.2d 67, 205 USPQ 594 (CCPA 1980); In re Swinehart, 439 F.2d 2109, 169 USPQ 226 (CCPA 1971).

Moreover, a newly discovered property does not render a compound unobvious, if (1) the claimed compound is structurally obvious from a prior art compound, (2) the claimed compound possesses the same property for which the prior art compounds were useful, and (3) the prior art compound in fact possesses the newly discovered property of the claimed compound. Monsanto Co. v. Rohm & Haas Co. (DC ED Pa 1970) 420 F2d 950, 164 U.S.P.Q 556.

For these reasons, Applicants' arguments are not persuasive, and the rejection of record is maintained.

### *Conclusion*

6. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date

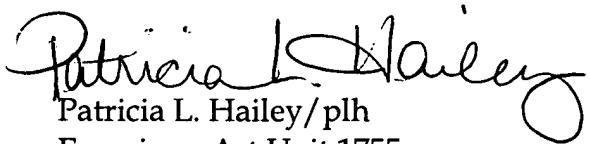
of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Patricia L. Hailey whose telephone number is (571) 272-1369. The examiner can normally be reached on Mondays-Fridays.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Jerry Lorengo, can be reached on (571) 272-1233. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the Group 1700 Receptionist, whose telephone number is (571) 272-1700.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

  
Patricia L. Hailey/plh  
Examiner, Art Unit 1755  
April 10, 2006

  
KARL GROUP  
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